NORTH GAUHATI COLLEGE

SESSIONAL EXAMINATION-2025

FYUGP SEMESTER-IV SUB: CHEMISTRY Major

Paper name: Theoretical Chemistry-1 Paper code: CHE0400304

TOTAL MARKS: 30 TIME: 1½ Hr

The figures in the margin indicate full marks for the questions.

1.	What are causes of failure of classical mechanics?	2
2.	What is black body radiation?	2
3.	Calculate the value of the Rydberg constant of hydrogen atom in eV.	3
4.	Calculate the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of an electron (m=9.1 $\times 10^{-31}$ kg) moving at 1% speeds to the de-Broglie wavelength of the de-Broglie wavelength	d of
	light. (h-6.626 \times 10-34 kgm ² S ⁻¹)	3
5.	What do you mean by zero point energy of a linear harmonic oscillator?	2
6.	What will happen if the walls of one-dimensional box are suddenly removed?	2
7.	Write an expression for Schrodinger wave equation. Explain the significance of the terms	
	involved.	3
8.	An electron in 1-D box of length 10 nm undergoes transition from the ground state to the	first
	excited state. Calculate wavelength of the photon absorbed.	3
9.	Using the principle of LCAO for the wave function for H_2^+ ion, obtain the normalized wa	ive
	functions for the bonding molecular orbitals.	2
10.	What are the postulates of valence bond theory?	3
11.	What type of intermolecular forces exist between the following substances:	3
	N2, HCl, NaCl	
12.	For a particle of 3-D box, calculate the degeneracy of energy level 14h ² /8ma ² .	2
